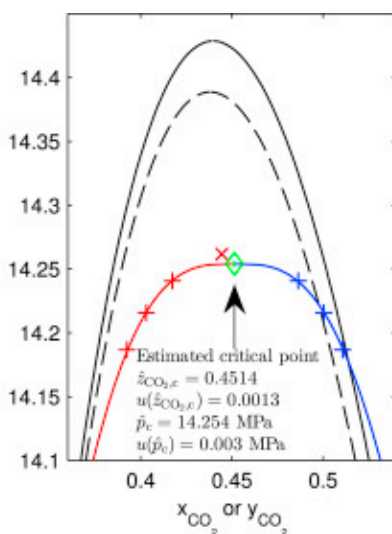
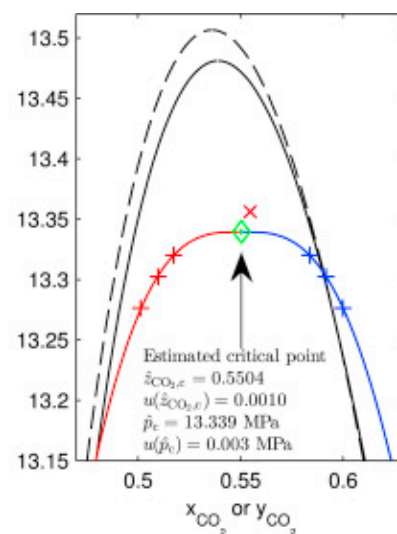


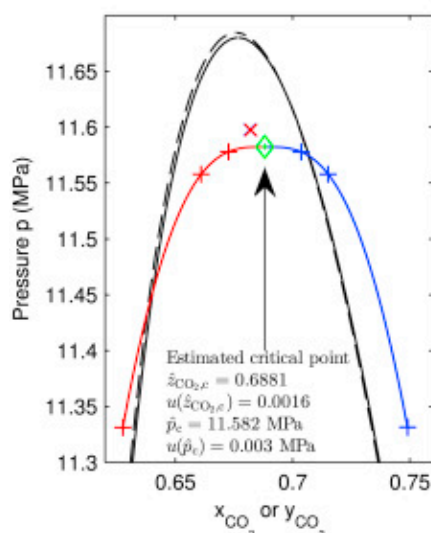
(a) 218.15 K. The figure shows only 6 of the 8 VLE points used for fitting the scaling law.



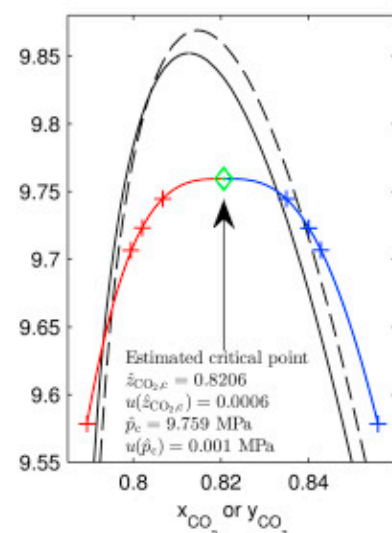
(b) 233.14 K. The figure shows only 6 of the 8 VLE points used for fitting the scaling law.



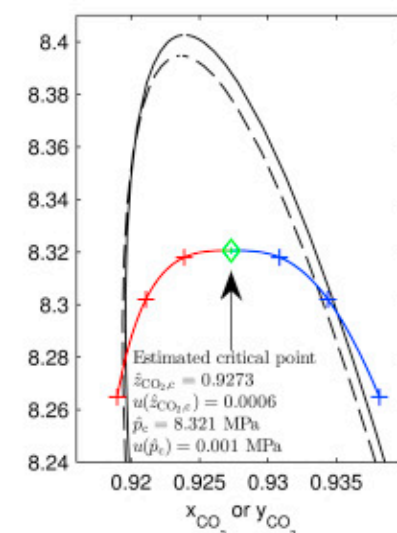
(c) 253.15 K. The figure shows only 6 of the 8 VLE points used for fitting the scaling law.



(d) 273.15 K.



(e) 288.14 K.



(f) 298.14 K.

- + Liquid phase measurements used for fitting scaling law
- + Vapor phase measurements used for fitting scaling law
- x Supercritical measurements taken using liquid phase sampler
- x Supercritical measurements taken using vapor phase sampler

- Scaling law bubble point curve
- Scaling law dew point curve
- - - PR-MC-WS-NRTL Case 1
- PR-MC-WS-NRTL Case 2

Previous article in issue: Vapor-Liquid Equilibrium Data Collection (Chemistry Data Series, Vol. 1, Part 1b). Von J. Gmehling, U. Onken und J. R. Rarey-Nies.

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The most complete collection of the azeotropic data was published by Gmehling et al. [6]. . [17,18]. Since Table 1 . the vapor-liquid equilibrium between THF and .

In thermodynamics and chemical engineering, the vaporliquid equilibrium .
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